

## 3-METHOXY-4,6-DIHYDROXYMORPHINANDIEN-7-ONE, AN ALKALOID FROM *CROTON BONPLANDIANUM*

K. P. TIWARI, R. N. CHOUDHARY and G. D. PANDEY

Department of Chemistry, University of Allahabad, Allahabad, India

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**Key Word Index**—*Croton bonplandianum*; Euphorbiaceae; 3-methoxy-4,6-dihydroxymorphinandien-7-one; structural determination.

**Abstract**—A new alkaloid, 3-methoxy-4,6-dihydroxymorphinandien-7-one, and norsinoacutine have been isolated from extracts of *Croton bonplandianum*.

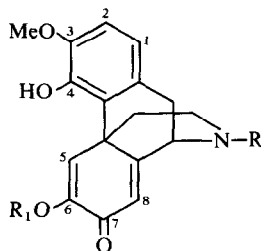
### INTRODUCTION

Several alkaloids [1–18] have been isolated from species of the genus *Croton*, but no work has been reported on *Croton bonplandianum*. A systematic chemical examination of this plant was, therefore, undertaken and the alkaloidal constituents are reported in the present communication.

### RESULTS AND DISCUSSION

The new alkaloid isolated is assigned the structure 3-methoxy-4,6-dihydroxymorphinandien-7-one (**1**)  $C_{17}H_{17}NO_4$  on the basis of following evidence. The presence of nitrogen as a secondary amine was suggested by the appearance of a band at  $3215\text{ cm}^{-1}$  in the IR spectrum [1]. The other IR bands at  $3400\text{ cm}^{-1}$  show the presence of an OH group and the bands at 1665, 1634 and  $1608\text{ cm}^{-1}$  indicate the presence of a dienone [1]. Furthermore the appearance of an UV absorption band at 238 nm confirms the presence of a cross conjugated system [2].

The compound (50 mg) when heated under reflux (2 hr) in MeOH with excess MeI in the presence of  $\text{Na}_2\text{CO}_3$ , yielded the *N*-*O*-trimethylmethiodide derivative (30 mg),  $C_{21}H_{26}INO_4 \cdot \text{Me}_2\text{CO}$ , mp  $163-5^\circ$  ( $\text{Me}_2\text{CO}$ ), which was found to be identical with *N*-*O*-dimethylnorsinoacutine methiodide [3], thereby suggesting that the new alkaloid differed from norsinoacutine  $C_{18}H_{19}NO_4$  (**2**) only in the presence of a OH group in place of a OMe group. The same methiodide is also obtained from the *N*-methyl-4,6-dihydroxy-3-methoxymorphinandienone [19] which confirms the presence of two OH groups and one OMe group.



The signals observed in the NMR spectrum at  $\delta$  6.56 (H, C-1), 6.83 (H, C-2), 7.63 (H, C-5), 6.15 (H, C-8) and 3.83 (3 H, OMe, C-3) are due to the presence of two aromatic protons, two olefinic protons and one OMe group, respectively [1]. The signal at  $\delta$  3.75, observed in norsinoacutine did not appear in the spectrum of **1**, showing the absence of a OMe group at C-4 [1].

The mass spectrum [4,20] of the compound showed a  $M^+$  at  $m/e$  299 and other prominent peaks at  $m/e$  284 ( $M^+ - \text{Me}$ ), 277 ( $M^+ - \text{CO}$ ), 256 ( $284 - \text{CO}$ ) and 164 which supported the proposed structure for the compound.

Another compound, mp  $113^\circ$ ,  $C_{18}H_{19}NO_4$  (**2**) was obtained from the alkaloidal mixture on column chromatography, which was identified as nonsinoacutine [5].

### EXPERIMENTAL

**Extraction and isolation of alkaloids.** The air dried and powdered plant (5 kg) was extracted with 2% HOAc soln. The extract (10 l.) was concd under red. pres. to 500 ml and basified with 25%  $\text{NH}_4\text{OH}$  soln. The basified extract was continuously shaken with  $\text{CHCl}_3$  for 2 hr and the organic layer separated. The solvent was removed *in vacuo* to yield a mixture of alkaloids. The crude alkaloidal mixture showed two spots on TLC, was adsorbed on an  $\text{Al}_2\text{O}_3$  column and eluted with  $\text{CHCl}_3$ -MeOH (19:1). The first eluate yielded an amorphous compound, which on crystallization from EtOAc at  $0^\circ$  yielded crystals of norsinoacutine [5] (310 mg), mp  $113^\circ$ . Successive elution with MeOH yielded another compound (145 mg), mp  $143^\circ$  (MeOH-EtOAc). Found: C, 65.19; H, 6.41; N, 3.59;  $C_{17}H_{17}NO_4$ . EtOAc requires C, 65.12; H, 6.46; N, 3.62%. UV  $\lambda_{\text{max}}^{\text{MeOH}}$  nm: 238; IR  $\nu_{\text{max}}^{\text{KBr}}$   $\text{cm}^{-1}$ : 3400, 3215, 1665, 1634, 1608; MS  $m/e$  (rel. int.): 299 ( $M^+$ ), 284, 277, 266, 256;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ )  $\delta$  3.83 (3 H), 6.15 (1 H), 6.56 (1 H), 6.83 (1 H), 7.63 (1 H).

**Preparation of methiodide derivative.** **1** (50 mg) when heated under reflux (2 hr) in MeOH with excess MeI in the presence of  $\text{Na}_2\text{CO}_3$  gave a product, which was adsorbed onto an  $\text{Al}_2\text{O}_3$  column. Elution with  $\text{Me}_2\text{CO}$ -MeOH (8:1) yielded the methiodide derivative (39 mg) which crystallized from  $\text{Me}_2\text{CO}$ , mp  $163-5^\circ$ . Found C, 53.1; H, 5.9;  $C_{21}H_{26}INO_4 \cdot \text{Me}_2\text{CO}$  requires C, 53.2; H, 6.0%. It was identical with *N*-*O*-dimethylnorsinoacutine methiodide [3].

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